

# Notes for Intro Notepacket

\* The first VARIABLE written is THE X  
**Graphical Relationships**

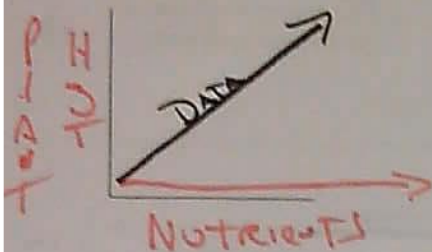
Fact(s) to memorize: 4-6

**A. Direct Relationship.**

As one variable increases, the other increases

Examples:

The more NUTRIENTS added  
The taller the PLANTS will  
Grow

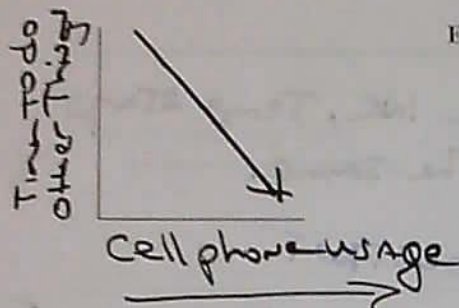


**B. Inverse Relationship**

As one variable increases, the other decreases

Examples:

Increase cell phone usage  
Time to do other things  
decreases



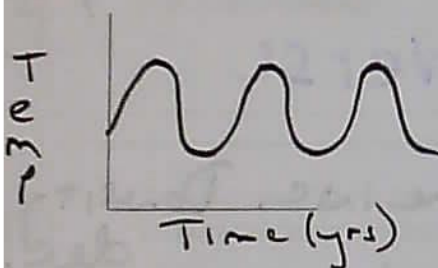
**C. Cyclic Relationship**

As one variable increases, the other will inc & dec  
in a PREDICTABLE pattern

Examples:

As the year inc.  
The Global Temp inc & dec.

\*Events that are cyclic are also PREDICTABLE!

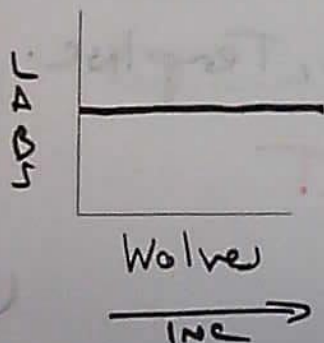


**- CONSTANT -**  
**D. No relationship.**

As one variable increases, the other Remain the same

Examples:

As the number of wolves increase  
The number of LAPS  
Remains the same

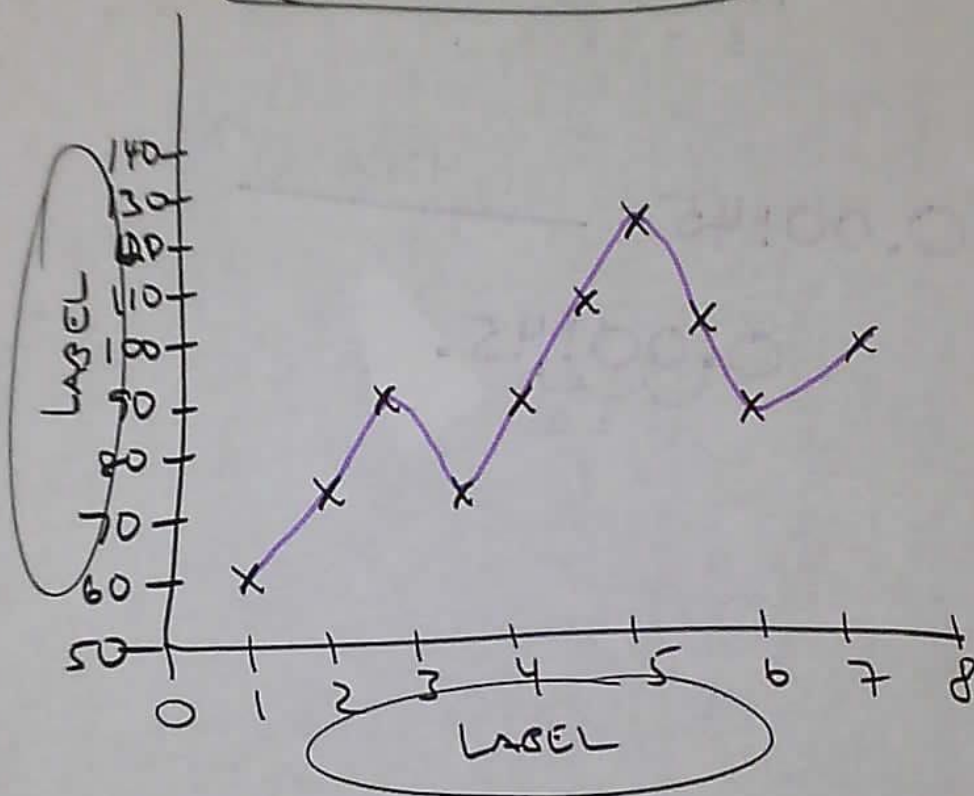


T A L E S  
Title Axis Axis Even Spacing  
Drawn Labeled

# How to Graph?

Title

Dependant Variable



Independent  
Variable

$$445,000 = \underline{4.45 \times 10^5}$$

$$\begin{array}{cccccc} 4 & . & 4 & 5 & 0 & 0 & 0 & . \\ \underbrace{\hspace{1.5em}} & & \underbrace{\hspace{1.5em}} & & \underbrace{\hspace{1.5em}} & & \underbrace{\hspace{1.5em}} & \\ 5 & & 4 & & 3 & & 2 & & 1 \end{array}$$

$$0.00145 = \underline{1.45 \times 10^{-3}}$$

$$\begin{array}{cccccc} 0 & . & 0 & 0 & 1 & . & 4 & 5 \\ \underbrace{\hspace{1.5em}} & & \underbrace{\hspace{1.5em}} & & \underbrace{\hspace{1.5em}} & & \underbrace{\hspace{1.5em}} & \\ 1 & & 2 & & 3 & & & \end{array}$$



Rate of Change

change in value (variable)  
over time

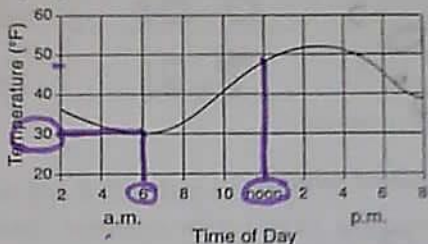
Formula for  
Rate of Change

$$ROC = \frac{\Delta \text{Value}}{\Delta \text{Time}}$$

Graphs

Visual rep. of data obtained  
in the experiment

- The temperature of water in a container was  $60^{\circ}\text{C}$ . Ten minutes later, the water temperature was  $35^{\circ}\text{C}$ . What was the rate of cooling of the water?  
 (1)  $25^{\circ}\text{C}/\text{min}$  (2)  $2.5^{\circ}\text{C}/\text{min}$  (3)  $35^{\circ}\text{C}/\text{min}$  (4)  $3.5^{\circ}\text{C}/\text{min}$
- The graph below shows the temperature readings for a day in April.

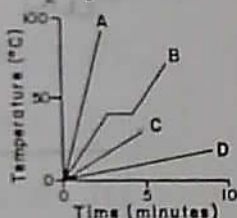


The average rate of temperature change, in Fahrenheit degrees per hour, between 6 a.m. and noon was

- (1)  $6^{\circ}/\text{hr}$  (2)  $8^{\circ}/\text{hr}$  (3)  $3^{\circ}/\text{hr}$  (4)  $18^{\circ}/\text{hr}$

$18^{\circ}\text{C}$   
 $\frac{40^{\circ}\text{C} - 30^{\circ}\text{C}}{6 \text{ hrs}}$

Use the graph to the right to answer questions 3 and 4. The graph represents the relationships between temperature and time as heat is added at a constant rate to equal masses of four substances labeled A, B, C, and D.



- The temperature of which substance increased the most rapidly? A
- Which substance has a change that is not at a constant rate? B
- Calculate the average daily rate of movement of the hurricane during the period from 3 p.m. August 24 to 3 p.m. August 28. The hurricane traveled 2,600 kilometers during this 4-day period.

$$\frac{2,600}{4 \text{ DAYS}}$$

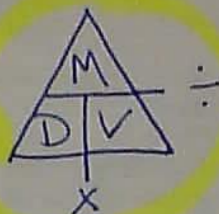
$$\underline{650 \text{ K/DAY}}$$

## Density of Matter

Density: Amount of mass in a given space

Formula:

$$\text{Density} = \frac{\text{Mass}}{\text{Volume}}$$

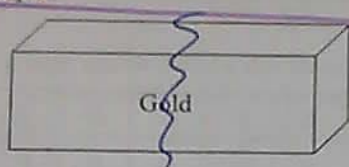


A. Density Properties:

~~Heat & Pressure can change Density.~~

~~\* The same material will have the same density \*  
NO matter the size.~~

Example:



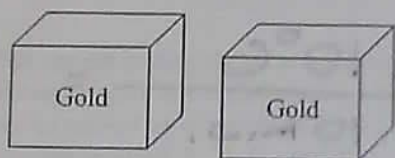
Mass = 162 g

Volume = 8.4 cm<sup>3</sup>

$$\text{Density} = \frac{\text{mass}}{\text{volume}} = \frac{\text{g}}{\text{cm}^3}$$

$$\text{Density} = 19.3 \text{ g/cm}^3$$

If you take that sample of gold and break it into two exact halves, the mass and volume is half of the original, but the density remains the same.



Mass = 81 g

Volume = 4.2 cm<sup>3</sup>

$$\text{Density} = \frac{\text{mass}}{\text{volume}} = \frac{\text{g}}{\text{cm}^3}$$

$$\text{Density} = 19.3 \text{ g/cm}^3$$

Practice questions:

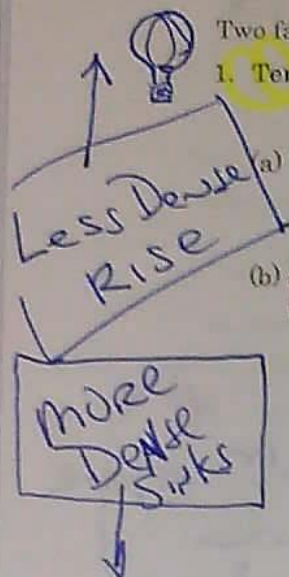
- If a wooden block were cut into eight identical pieces, the density of each piece compared to the density of the original block would be
  - less
  - greater
  - the same
- Under the same conditions of temperature and pressure, three different samples of the same uniform substance would have the same
  - shape
  - density
  - mass
  - volume



B. Change in Density:

Two factors that do effect density are Heat and Pressure

1. Temperature As temperature increases, molecules begin to move a part (expand), which means the volume increases.



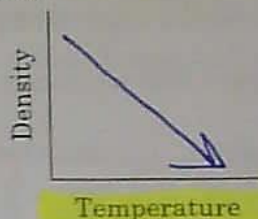
(a) Warm air Rises because it is Less dense.

An example would be a HOT AIR balloon

(b) State the relationship between temperature and density.

As Temp INC. Density decrease

(c) Draw the relationship between temperature and density in the graph below.



Phases of Matter:

(a) Most materials have their greatest density as a SOLID

The exception is water, because water expands when it freezes.

(b) Water is at its greatest density at a temperature of 4 °C  
The density of water is 1 g/ml

(c) If an object floats on water, it is less dense than the water.  
If an object sinks in water, it is more dense than the water.

2. Pressure: When pressure is added, it causes the material to become smaller (compress), volume decreases.

(a) State the relationship between pressure and density.

As pressure INCREASES Density INCREASES

(b) Draw the relationship between pressure and density in the graph below.

